

# Accuracy of density functional theory methods for the calculation of magnetic exchange couplings in binuclear iron(III) complexes



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## ABSTRACT

We assess the performance of popular hybrid density functional approximations for the prediction of magnetic exchange couplings ( $J$ ) in Fe<sup>III</sup> complexes. To this end, eleven dinuclear oxo-bridged iron(III) complexes with accurately determined  $J$  values were chosen to cover cases of weak and strong antiferromagnetic couplings ranging from  $-6\text{ cm}^{-1}$  to  $-132\text{ cm}^{-1}$ . Seven representative exchange-correlation functionals ranging from the hybrids PBEh and B3LYP to meta hybrids (M06, M062X), range separated functionals (HSE, LC- $\omega$ PBE), and LSDA with 30% of HF exchange were used for this assessment. These functionals have been suggested as good performers for this particular property in transition metal complexes and hence this assessment aims to narrow the interval of confidence for the particular case of Fe complexes. In addition, we examined the performance of an alternative method based on differential local spin rotations to calculate magnetic exchange coupling parameters in these complexes. Our results show that, although some of these functionals perform on par, the M06 and HSE functionals yield magnetic exchange couplings in closer agreement with experimental data, with mean absolute percentage errors of 8.6 and 9.2%, respectively.

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## 1. Introduction

Transition metal (TM) complexes show distinct magnetic properties due to the presence of exchange interactions between metal centers [1,2], spin-orbit coupling [3], and their variable coordination number [4] and oxidation state. Their characteristic low dimensionality and their capability to exhibit different spin ground states as a result of competing magnetic interactions make these complexes of significant interest as potential molecular magnets (MM) [5], spintronics devices [6], memory chips [7], and quantum computing units [8]. Moreover, they are widely studied by biochemists due to their crucial role in several biochemical processes [9].

Theoretical and experimental studies of the magnetic properties of transition metal complexes have been directed towards designing novel molecular magnets with desired properties and performance [5,10–12,13–17]. Many of these studies employ magneto-structural correlations to predict and understand the origin of the magnetic interactions. This has been accomplished

experimentally by magnetochemical studies, while from the theoretical side, several electronic structure methods have been employed, broadly divided into wave function-based post-Hartree-Fock [18–21], and Kohn-Sham (KS) density functional theory (DFT) [22] methods. While the former provide a systematically improvable and generally accurate route to calculating magnetic properties, calculations on medium- and large-size TM complexes using these methods are unaffordable. DFT methods, on the other hand, provide a widely-used and computationally accessible alternative to wave function methods that have found routine use as a tool for the prediction of electronic properties of materials ranging from molecules to solids. Despite the fact that DFT is an exact theory in principle, in practice the choice of the approximate exchange correlation energy functional ( $E_{xc}$ ) [23] determines the accuracy of the calculations. The practical realization of DFT started with the development of local density approximation (LDA) [24,25] where  $E_{xc}$  only depends upon the local electron density. The LDA [25] performs reasonably well for some systems and properties but fails for many others. The generalized gradient approximation (GGA) [26,27] was introduced by incorporating gradients of the electron density, and although it improves upon the LDA in several respects, it still falls short for properties such as dissociation energies,

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energy gaps, and magnetic properties [26]. Magnetic exchange couplings calculated with LDA and GGA in diverse TM complexes are on average 6 to 11 times larger than reference experimental values [28].

With the advent of hybrid functionals, DFT has gained more accuracy for several molecular properties and reached a larger variety of systems. These functionals are widely implemented in DFT codes and their accuracy for the prediction of structural and electronic properties of materials is well documented [29–31,28]. Although the performance of hybrid density functionals for magnetic exchange couplings has been analyzed in the past, deviations from accurate experimental data are typically larger than for other properties [32–34]. With the significant advances in the quality of density functionals in the past years, there is always interest regarding their reliability for the calculation of magnetic properties [35,27,36–38,29,39–41,31,42,43,30,44,45]. For instance, Phillips et al. [28] have shown statistical errors of about 40% in a set of 14 bimetallic homonuclear transition metal complexes containing Fe, Mn, Cu, V metals. Since the interactions that take place in these complexes are very different (giving a range of exchange couplings from  $-382 \text{ cm}^{-1}$  to  $+114 \text{ cm}^{-1}$ ), it is reasonable to wonder what is the performance of hybrid density functionals in a set of same-metal homonuclear TM complexes. This type of analysis can provide a more focused appraisal of practical use. In this work, we assess the performance of representative density functionals for the prediction of magnetic exchange couplings involving  $\text{Fe}^{\text{III}}$  centers. This has two goals: to assess the performance of these functionals and also to determine the expected errors for this specific type of exchange couplings. To this end, we selected a set of 11 dinuclear oxo-bridged  $\text{Fe}^{\text{III}}$  complexes with accurately determined experimental  $J$  couplings. These complexes were chosen so that they can be considered as representative of typical oxo-bridged  $\text{Fe}^{\text{III}}-\text{Fe}^{\text{III}}$  moieties and hence the conclusions can be extrapolated to larger  $\text{Fe}^{\text{III}}$  complexes. It should be mentioned that this assessment validates the ability of the selected DFT approximations to reproduce  $J$  couplings, and not the errors in the resulting densities associated with these functionals [46], which could affect other magnetic properties.

## 2. Computational details

For most TM complexes, magnetic interactions are typically modeled using the Heisenberg-Dirac-van-Vleck Hamiltonian [47] in terms of the magnetic exchange coupling parameter  $J_{ab}$  defined by

$$H_{\text{HDVV}} = -2 \sum_{a < b} J_{ab} \vec{S}_a \cdot \vec{S}_b, \quad (1)$$

where  $S_a$  and  $S_b$  are the (localized) spin moments on magnetic centers  $a$  and  $b$ , respectively. For a dinuclear system,  $J_{ab}$  can be extracted from DFT calculations using high-spin (HS) and broken symmetry (BS) DFT reference states using

$$J_{ab} = \frac{E_{\text{BS}} - E_{\text{HS}}}{4S_a S_b}. \quad (2)$$

This approach is referred to as BS-DFT and its accuracy depends on the choice of the approximate  $E_{\text{xc}}$  employed [48,49].

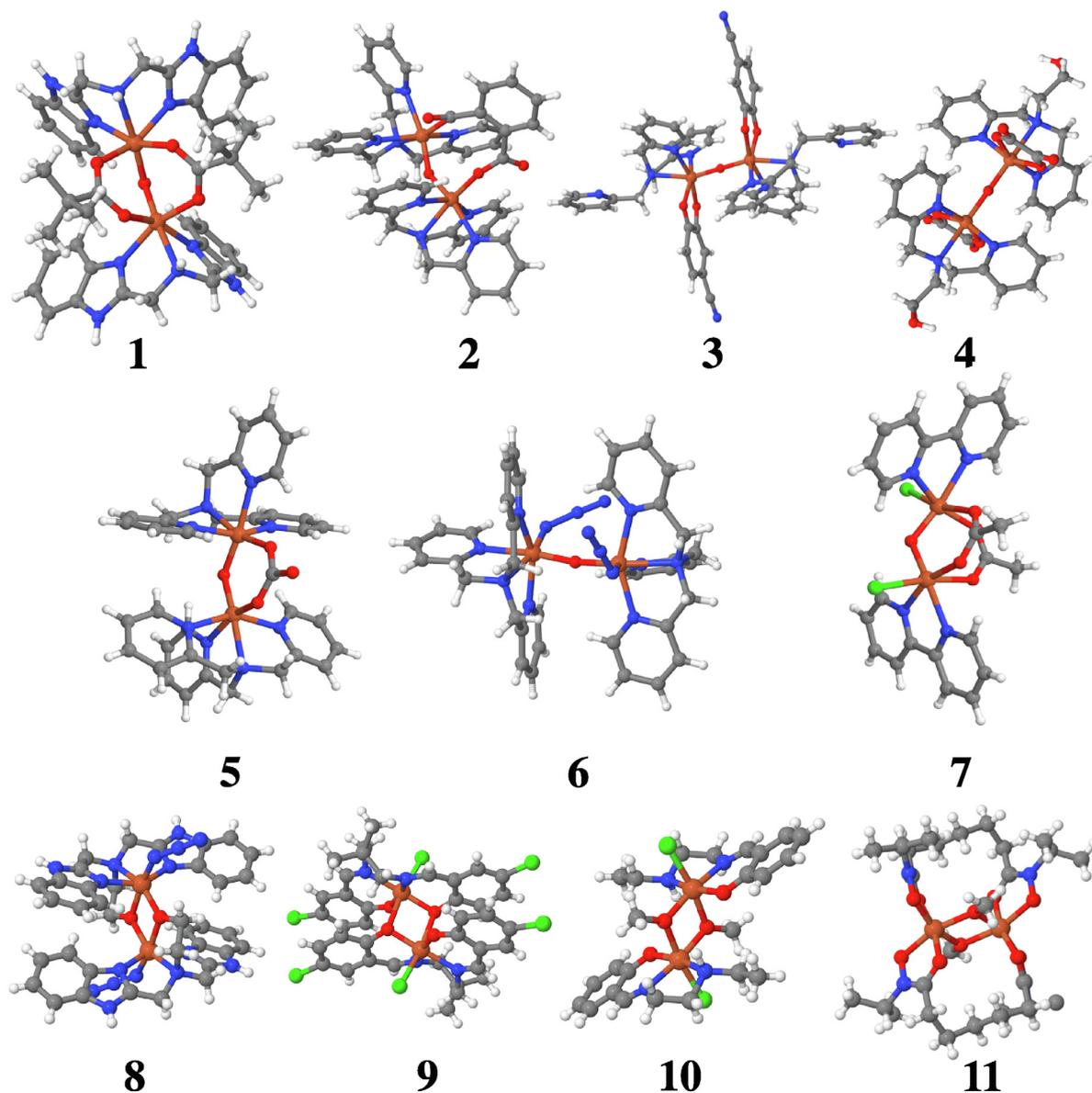
In addition, in this work we have investigated the performance of a recently proposed method for calculating magnetic exchange coupling parameters starting from a particular spin configuration, either HS or BS. This method is based on a generalized perturbative approach within DFT where magnetic exchange coupling parameters are given by the analytic second derivative of the energy with respect to local spin rotations.

$$J = \frac{1}{2S_1 S_2} \frac{d^2}{d\theta^2} \langle \hat{H}_{\text{HDVV}} \rangle \Big|_{\theta=0}. \quad (3)$$

Within this method, the relative angle between two spins is used as a constraint, while a torque on the local spin is used as a small perturbation. We refer readers to Refs. [50,51] for details. Other related methods for the calculation of exchange couplings in molecules that utilize approximate differential local spin rotations based on Green's functions have been proposed [52,53]. Within this approach, the explicit determination of different spin solutions is avoided and, instead, the noncollinear coupled-perturbed KS equations need to be solved to determine the second derivative in Eq. (3). We refer to this method as BB for black-box. One of the drawbacks of the BB method is its difficulty to converge self-consistently when using a density functional that includes a GGA component. For this reason we have adopted the local LSDA exchange-correlation functional with 30% of HF exchange in it (herein called LSDA30) for all calculations using this method.

Fig. 1 shows the molecular complexes used in this work. These complexes have been investigated by single-crystal X-ray diffraction and magnetochemical studies. Structural parameters for the complexes were taken from the corresponding crystal structures. All the complexes considered here are dinuclear  $\text{Fe}^{\text{III}}$ , and most have a single bridging oxido ( $\text{O}^{2-}$ ) ion with either zero, one, or two additional bridging ligands to provide a variety of structural types, bridging Fe-O bond lengths and Fe-O-Fe angles, and resulting relatively strong exchange couplings (complexes 1-7). In addition, we include four complexes that are doubly-bridged either by two methoxo ( $\text{MeO}^-$ ) ions (complexes 10 and 11), or alkoxo- or phenoxo-units within more complicated bridging ligands (complexes 8 and 9, respectively). These latter four complexes are known experimentally to exhibit much weaker exchange couplings than oxido-bridged complexes, ensuring a wide range of coupling strengths for our computational study. In all of these complexes, the Fe atoms are in a +3 oxidation state with five unpaired electrons in  $3d$  orbitals, and the magnetization originates mainly due to the interaction between magnetic moments at the metal centers with negligible contribution from other atoms in the ligands attached to them. For consistency, all  $J$  coupling calculations in this work were carried out using each experimentally determined crystal structure. All complexes considered have experimental antiferromagnetic exchange coupling parameters ranging from  $-6 \text{ cm}^{-1}$  to  $-132 \text{ cm}^{-1}$ , and all the HS states have an effective spin multiplicity of 11 while the BS states have a multiplicity of 1.

For all calculations we have used the Gaussian09 suite of programs [54] to evaluate the total energy for the HS and BS spin states with seven exchange correlation functionals, starting from the local hybrid LSDA30 to the widely used hybrid functionals PBEh [27,36,37] and B3LYP [26,55,35,56,57], to the meta hybrid functionals M06, M062X [30], and finally range separated functionals HSE [38,29,39–41] and LC- $\omega$ PBE [31,42,43]. LSDA30 is the hybrid of LSDA with 30% HF exchange which has been previously described. PBEh [28,36,37] is the hybrid functional of Perdew, Burke and Ernzerhof (also known as PBE0 or PBE1PBE in the literature) with 75% of its exchange energy determined by PBE exchange and 25% from Hartree-Fock exchange, with correlation from the PBE functional. The meta-GGA functionals M06 and M062X admix 27% and 54% HF exchange, respectively, with DFT exchange [30]. We employed the 2006 version of the hybrid HSE [38] (also known as HSE06 in the literature; keyword HSEH1PBE in Gaussian), which admixes short-range screened Hartree-Fock exchange in the functional realization. LC- $\omega$ PBE [58] is a long-range corrected GGA functional which partitions exchange interactions into short and long range terms where short range-exchange interactions are described with PBE exchange while long range



**Fig. 1.** Crystal structures of the complexes considered in this work with their CCDC names (1) JEJVEZ, (2) VIGHEY, (3) FAJPUD, (4) FAJQEO, (5) VIGGUN, (6) FAJQAK, (7) VABMUG, (8) ABIZOA, (9) ELISAU, (10) BAZCOV, (11) BOSMOL10. Here, the Fe are represented by dark brown balls, oxygen by red, nitrogen by blue, chlorine by green, carbon by gray, and hydrogen by white. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

interactions are treated by HF exchange. Convergence to the target magnetic state was confirmed by a minimum basis set (MBS) Mulliken spin population analysis [59]. All calculations were converged to  $10^{-6}$  ha =  $0.2$  cm $^{-1}$  in energy and  $10^{-8}$  RMS change in the density matrix.

### 3. Results and discussion

We first checked for a suitable basis set for our calculations by examining the basis set dependence of  $J$  for the particular complex **2** as a representative case with the choice of M06 exchange-correlation functional. Five different basis set were employed for this purpose: (i) 6-31G\*\* on all atoms, (ii) 6-311G\*\* on all atoms, (iii) 6-311+G\* on all atoms, (iv) 6-311+G\*\* on all atoms, and (v) 6-311+G\*\* for Fe and 6-31G\*\* for C, N, O, H atoms. The calculated  $J$  couplings for these basis sets are  $-106.7$ ,  $-111.0$ ,  $-109.0$ ,  $-109.0$ , and  $-109.6$  cm $^{-1}$  for the basis set (i), (ii), (iii), (iv), and

(v), respectively. These results show that although all these basis sets give comparable  $J$  couplings, option (v) provides a reasonable compromise between quality and computational effort. All further calculations in this work were performed with this choice of basis set.

In Table 1 we show the calculated exchange couplings for all the complexes using the different exchange-correlation functionals along with the experimentally determined couplings. For all the complexes, HSE and PBEh give fairly similar results, with HSE being slightly closer to experimental values compared to the PBEh couplings. Although LC- $\omega$ PBE performs very well in a number of applications such as energy barriers, bond lengths, thermochemistry, and long-range charge transfer process for molecular systems [31], our results show that this long-range corrected hybrid functional generally overestimates Fe-Fe  $J$  couplings. On the other hand,  $J$  couplings calculated with the meta GGA hybrid functional M06, which admixes 27% of Hartree Fock exchange, are in good agreement with experiments. This contrasts with our early

**Table 1**  
Magnetic exchange couplings ( $\text{cm}^{-1}$ ) for the dinuclear iron complexes considered in this work using the BS approach. For the LSDA30, couplings using both the BS approach and the BB method are shown.

Complex	Functional								Exp.
	HSE	PBEh	B3LYP	LC- $\omega$ PBE	M06	M062X	LSDA30		
							BS	BB	
<b>1</b>	-112.6	-109.8	-132.2	-121.5	-116.9	-68.3	-99.4	-106.6	-116.0 <sup>a</sup>
<b>2</b>	-112.0	-106.5	-124.9	-119.4	-109.6	-69.1	-96.8	-106.0	-110.0 <sup>b</sup>
<b>3</b>	-95.4	-95.0	-106.2	-112.9	-94.5	-67.9	-86.3	-95.4	-78.0 <sup>c</sup>
<b>4</b>	-97.1	-95.7	-109.9	-115.4	-97.6	-64.5	-86.8	-93.1	-98.0 <sup>c</sup>
<b>5</b>	-100.0	-98.0	-110.2	-102.7	-103.9	-59.6	-84.8	-94.4	-108.0 <sup>b</sup>
<b>6</b>	-103.6	-102.0	-117.1	-117.6	-104.3	-69.8	-93.0	-99.9	-90.0 <sup>c</sup>
<b>7</b>	-113.6	-111.1	-132.3	-124.8	-116.4	-70.3	-100.3	-111.0	-132.0 <sup>d</sup>
<b>8</b>	-18.7	-18.6	-22.2	-20.3	-20.3	-9.8	-16.3	-15.5	-21.3 <sup>e</sup>
<b>9</b>	-5.9	-5.8	-7.6	-4.9	-6.7	-3.2	-4.9	-4.5	-6.4 <sup>f</sup>
<b>10</b>	-15.0	-14.8	-18.2	-14.9	-15.4	-7.2	-12.3	-11.6	-13.65 <sup>g</sup>
<b>11</b>	-11.0	-10.9	-13.3	-11.8	-12.0	-5.2	-8.6	-7.8	-10.9 <sup>h</sup>

<sup>a</sup> Taken from Ref. [65].

<sup>b</sup> Taken from Ref. [66].

<sup>c</sup> Taken from Ref. [67].

<sup>d</sup> Taken from Ref. [68].

<sup>e</sup> Taken from Ref. [69].

<sup>f</sup> Taken from Ref. [70].

<sup>g</sup> Taken from Ref. [71].

<sup>h</sup> Taken from Ref. [72].

findings for a set of 6 heterodinuclear TM complexes where HSE performs much better than M06 [51]. Our results with the M062X functional (containing 54% of Hartree Fock exchange) show that in this case, increasing the Hartree Fock admixture beyond a certain limit has an adverse effect on the value of the exchange coupling parameter, leading to a severe underestimation of the strength of the antiferromagnetic interaction.

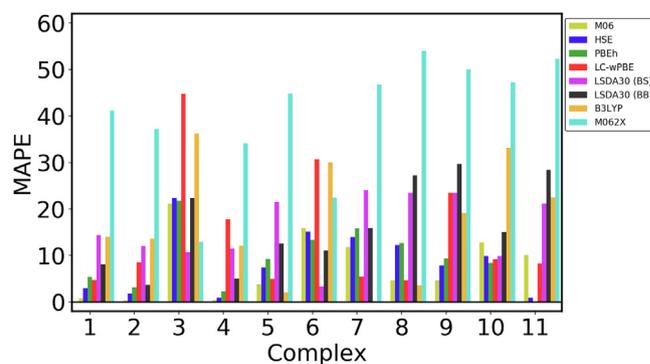
It should be mentioned that another popular alternative to the BS-DFT approach for mapping DFT energies into  $J$  couplings is based on Yahaguchi's formula, which effectively replaces  $4S_a S_b$  by  $\langle S^2 \rangle_{HS} - \langle S^2 \rangle_{BS}$  in Eq. (2) [60,61]. For our calculations,  $S_a = S_b = 5/2$ , and  $\langle S^2 \rangle_{HS} \approx 30$  and  $\langle S^2 \rangle_{BS} \approx 5$  in all cases due to the large localization of the spins at the Fe centers, which makes both approaches yield very similar values, typically within 1%.

We show in Table 1  $J$  couplings calculated with LSDA30 using both the energy-differences BS-DFT approach and the BB method. For this particular functional, calculated  $J$  couplings using the BS-DFT approach are systematically underestimated (in absolute value) when compared to their BB counterparts. This can be understood intuitively, recalling that  $J$  couplings obtained with the BB approach are formally identical to the energy-differences couplings only in the case of an ideal (perfectly localized) HDVV system, where the curvature of the energy as a function of the inter-spin angle is perfectly quadratic [50]. In real complexes, a small deviation of the BB couplings (calculated at the HS or BS states) from the BS-DFT results can be expected. Our results show that in this particular case, the choice of the methodology could have an effect comparable to the choice of the XC functional. However, the advantage of the BB method is highlighted in the case of large multicenter systems, where it requires only one spin state to calculate  $J$ , and hence it is free from typical problems arising from the BS-DFT approach, such as convergence difficulties for some BS solutions or convergence to undesired solutions.

For all the calculations performed in this work, the slight deviation between the experimental and theoretical values may be also partly attributed, among other reasons, to the fact that calculations are performed in gas phase and zero K, in contrast to experimental crystal phase and temperature dependent experiments. In this regard, our assessment gives an estimation of the errors originating

in the entire protocol for evaluating exchange couplings, and not only in the exchange-correlation functional chosen. The absolute percentage error of calculated  $J$  couplings with respect to experimental results is shown as a bar plot in Fig. 2.

We have identified a few weakly ferromagnetically-coupled dinuclear azido-bridged  $\text{Fe}_2^{\text{III}}$  complexes in the CCDC database [62–64] with  $J$  couplings between +0.8 and +2.4  $\text{cm}^{-1}$ . For the complex with the largest ferromagnetic coupling (CCSD name RIJLAX; Ref. [62]) we found that the calculated  $J$  is ferromagnetic, with



**Fig. 2.** Absolute percentage error of the calculated  $J$  couplings with respect to the experimental results for each complex.

**Table 2**  
Standard error deviation (in  $\text{cm}^{-1}$ ), mean percentage error (MPE), and mean absolute percentage error (MAPE) for different exchange-correlation functionals.

Method	Standard error deviation	MPE	MAPE
HSE	9.6	0.8	9.2
PBEh	10.1	-0.9	9.6
B3LYP	10.5	17.2	17.2
LC- $\omega$ PBE	13.7	8.1	15.8
M06	8.5	4.4	8.6
M062X	20.8	-40.2	40.2
LSDA30 (BS)	11.9	-13.3	16.1
LSDA30 (BB)	10.4	-10.4	17.1

**Table 3**

Minimum basis set Mulliken (MBS) atomic spin populations for the high spin (HS) and broken symmetry (BS) states for different dinuclear iron complexes and density functional approximations.

Complex	Functional									
	HSE		PBEh		LC- $\omega$ PBE		M06		M062X	
	HS	BS	HS	BS	HS	BS	HS	BS	HS	BS
<b>1</b>	4.40	-4.37	4.41	-4.38	4.42	-4.38	4.41	-4.38	4.52	-4.51
<b>2</b>	4.39	-4.34	4.40	-4.35	4.41	-4.35	4.40	-4.35	4.52	-4.49
<b>3</b>	4.36	-4.32	4.37	-4.32	4.39	-4.33	4.38	-4.32	4.50	-4.47
<b>4</b>	4.38	-4.33	4.39	-4.34	4.41	-4.35	4.38	-4.33	4.51	-4.48
<b>5</b>	4.36	-4.34	4.37	-4.34	4.39	-4.35	4.36	-4.33	4.49	-4.44
<b>6</b>	4.35	-4.29	4.36	-4.29	4.36	-4.30	4.34	-4.29	4.48	-4.45
<b>7</b>	4.37	-4.34	4.37	-4.34	4.39	-4.35	4.37	-4.34	4.49	-4.48
<b>8</b>	4.34	-4.34	4.34	-4.34	4.36	-4.36	4.34	-4.34	4.50	-4.50
<b>9</b>	4.37	-4.37	4.37	-4.37	4.39	-4.39	4.37	-4.37	4.51	-4.51
<b>10</b>	4.38	-4.38	4.37	-4.37	4.39	-4.39	4.37	-4.37	4.52	-4.52
<b>11</b>	4.41	-4.41	4.41	-4.41	4.43	-4.43	4.41	-4.41	4.54	-4.54

values of about  $6 \text{ cm}^{-1}$  for B3LYP and PBEh. This indicates that, even though these complexes fall outside the oxo-bridged category, this same protocol for evaluating exchange couplings can also predict ferromagnetic couplings.

We have also evaluated the standard deviation, mean percentage error and mean absolute percentage error in the calculated exchange couplings for each functional and show it in Table 2. Our results indicate that the M06 functional (with a standard error deviation and mean absolute percentage error of  $8.5 \text{ cm}^{-1}$  and 8.6%, respectively) is the most accurate functional for the calculation of Fe-Fe exchange couplings, compared to the other functionals considered, followed closely by HSE and PBEh, while M062X gives the largest error. The LSDA30 functional in combination with the BS energy-differences approach shows a relatively large standard deviation and mean absolute error, while this same functional performs slightly better when used with our BB method. This can be likely attributed to a compensation of factors: on the one hand, the LSDA30 tends to underestimate the coupling in absolute value (Table 1), while the second derivative of the energy with respect to local spin rotations given by the BB method is slightly larger than that of an ideal HDVV system.

We analyzed the atomic spin population dependence of each HS and BS state on the choice of exchange-correlation functionals using the minimum basis set Mulliken (MBS) population analysis. Our calculations show a small deviation from the nominal spin values of 5.0 at the Fe<sup>III</sup> centers, as shown in Table 3 for some selected functionals. This is expected as a result of a delocalization of the unpaired 3d electrons to the ligands. We obtained MBS spin populations at the Fe centers between 4.3 and 4.5 (in absolute value) in all cases, with larger values corresponding to M062X, which can be attributed to the increased localization produced by larger HFX contribution (54%) compared with other functionals in this work [73]. This larger localization can be related to the poor performance of M062X for this particular case.

#### 4. Conclusions

In this work we assessed the performance of seven popular hybrid density functional approximations for the calculation of Fe-Fe magnetic exchange coupling constants by comparing with reliable experimental values. Using the broken-symmetry energy-differences approach, our results show that the M06 exchange-correlation functional can predict Fe-Fe  $J$  couplings with a mean absolute percentage error (MAPE) of 8.6%, while HSE and PBEh yield slightly larger, but still small, MAPEs of 9.2% and 9.6%, respectively. We have also included in our assessment the BB method in combination with the hybrid LSDA30 functional and

show that in this case we get as similar MAPE as can be expected. Our analysis provides an estimate of the error in the prediction of magnetic exchange coupling constants in the density functional theory framework, particularly hybrid functionals, for the case of oxo-bridged Fe<sup>III</sup> complexes.

Hybrid functionals differ from the corresponding non-hybrids by including some fraction of exact exchange in the description of the electron-electron interaction. One result of this is to lessen the magnitude of one-electron self-interaction error (SIE) that arises from approximating the exact exchange-correlation functional. SIE can have a substantial impact on the performance of a non-hybrid functional. In recent work [74], some of us demonstrated that removing self-interaction effects from LSDA directly, by applying the Fermi-Löwdin orbital self-interaction correction (FLO-SIC), significantly improves the description of magnetic exchange couplings in a set of organic biradicals and in a Cu-based bimetallic complex. Removing self-interaction results in greater localization of unpaired electrons. This correctly increases the size of the calculated couplings for the radicals, while shrinking it for the bimetallic complex. The FLO-SIC-LDA results are competitive with PBEh results overall, and better in many cases. How FLO-SIC would perform for high-spin systems like the Fe<sup>III</sup> complexes described in this paper is an open question that we are currently addressing. The present results offer useful data for any electronic structure method.

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